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## acid dissociation constant wikipedia

in chemistry an acid dissociation constant also known as acidity constant or acid ionization constant denoted is a quantitative measure of the strength of an acid in solution it is the equilibrium constant for a chemical reaction known as dissociation in the context of acid base reactions the chemical species ha is an acid that dissociates into a the

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*alkali metal wikipedia*

the alkali metals consist of the chemical elements lithium li sodium na potassium k rubidium rb caesium cs and francium fr together with hydrogen they constitute group 1 which lies in the s block of the periodic table all alkali metals have their outermost electron in an s orbital this shared electron configuration results in their having very similar characteristic

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## history of science wikipedia

the history of science covers the development of science from ancient times to the present it encompasses all three major branches of science natural social and formal the earliest roots of science can be traced to ancient egypt and mesopotamia in around 3000 to 1200 bce their contributions to mathematics astronomy and medicine entered and shaped greek natural

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## inorganic chemistry 4th edition catherine housecroft

the present paper describes a new tripodal ligand containing imidazole and pyridine arms and its first cis ruiii l cl 2 clo4 complex 1 the crystal structure of 1 shows ruiii in a distorted octahedral geometry in which two chloride ions cis positioned to each other are coordinated besides the four nitrogen atoms from the tetradentate ligand l

*ionic compound wikipedia*

in chemistry an ionic compound is a chemical compound composed of ions held together by electrostatic forces termed ionic bonding the compound is neutral overall but consists of positively charged ions called cations and negatively charged ions called anions these can be simple ions such as the sodium na and chloride cl in sodium chloride or polyatomic

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password requirements 6 to 30 characters long ascii characters only characters found on a standard us keyboard must contain at least 4 different symbols

[metalloid wikipedia](#)

a metalloid is a type of chemical element which has a preponderance of properties in between or that are a mixture of those of metals and nonmetals there is no standard definition of a metalloid and no complete agreement on which elements are metalloids despite the lack of specificity the term remains in use in the literature of chemistry the six commonly recognised metalloids are

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detailed picture of the atom then builds toward

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**nonmetal wikipedia**

in chemistry a nonmetal is a chemical element that generally lacks a predominance of metallic properties they range from colorless gases like hydrogen to shiny solids like carbon as graphite the electrons in nonmetals behave differently from those in metals with some exceptions those in nonmetals are fixed in place resulting in nonmetals usually being poor